Research Article

Biological screening of some novel pyrimidine compounds

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Some novel pyrimidine compounds have been synthesized and their structural confirmation was done by different spectroscopic techniques such as FT-IR, NMR and MS. The biological screening of all the synthesized compounds is done in DMSO using agar well diffusion method. For the biological study, different Gram positive and Gram negative bacterial and fungal strains are used. It is observed that inhibition depends on strain, solvent and structure of compounds. In the present work, all the compounds were screened in DMSO solvent. The synthesized compounds have different moieties as well as substitutions. So, different strains are affected differently by different compounds.

Keywords: Pyrimidine compounds, DMSO, antimicrobial activity, agar well diffusion method

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Introduction

The extensive use of antibiotics has led to the appearance of multidrug resistant microbial pathogens [1]. In recent years, multiple drug resistance has developed due to indiscriminate use of existing antimicrobial drugs in the treatment of infectious diseases. Further, antibiotics are sometimes associated with adverse effects on the host-like hypersensitivity. Therefore, there is a need to develop alternative antimicrobials drugs for the treatment of infectious diseases from other sources.

The biological activity spectrum of a compound represents the pharmacological effects, physiological and biochemical mechanisms of action, specific toxicity that can be revealed in compound's interaction with biological system. Further, it describes the intrinsic properties of the compound, which depends on its structure.

Pyrimidines are always an attraction point for researchers because of its efficiency towards various pharmacological usages. These compounds are known to possess various biological activities [2-10]. Literature survey shows that various fused pyrimidine derivatives are known to exhibit anti-tubercular [11, 12], anti-proliferative [13, 14], anti-HIV [15, 16], anti-microbial [17], anti-analgesic [18], anti-inflammatory [19] and anti-malarial [20] activities. Compounds containing imidazo [2, 1-b] thiazole derivatives are also of great interest among medicinal

chemists as these compounds have also been reported for a wide spectrum of other biological properties [21-26].

In the present paper, some novel pyrimidines compounds have been synthesized. The antimicrobial activities of synthesized compounds have been screened against some bacterial (both Gram positive and Gram negative) and fungal strains in DMSO. The results are reported as minimum inhibitory concentrations (MIC) and minimum bactericidal concentration (MBC) for all the synthesized compounds.

Materials and Methods

Drug synthesis

Synthesis of 2, 4-disubstituted pyrimidine derivatives (BKD-1 to BKD-12): In *n*-butanol, equimolar mixture of 2, 4-dichloropyrimidine (DCP), 4-((1*H*-1, 2, 4-triazol-1-yl) methyl) aniline (TMA) and 0.012 mole of N, N-diisopropyl ethyl amine was refluxed for 3 hr. The completion of reaction was confirmed by analytical thin layer chromatography (TLC) using as a 9.6 : 0.4 dichloromethane : methanol mobile phase. After completion of reaction, reaction mixture was cooled. The resulting solid was filtered, washed with cold water and dried under vacuum to give crude product.

This resulting product was refluxed for 3 hr with ethanolic solution of different aromatic amines (0.011 mol)

solid was filtered, washed with cold ethanol and dried under vacuum to give crude product. The obtained crude

product was purified by tituration with diethyl ether.

using glacial acetic acid as catalyst. The completion of reaction was monitored using TLC (100% ethyl acetate + NH³ atmosphere as a mobile phase). After completion of reaction, the reaction mixture was cooled and the resulting

The reaction scheme is:

Synthesis of 2, 4-disubstituted pyrimidine derivatives 2,4-dichloropyrimidine (DCP), 1-Naphthol (NTL) and 0.015 mole of Potassium carbonate (K_2CO_3) in DMF was refluxed for 4 hr. The completion of reaction was confirmed by analytical thin layer chromatography (TLC) using (7:3–Hexane: Ethyl acetate) as mobile phase. After completion of reaction, the reaction mixture was cooled and the resulting solid was filtered, washed with cold water and dried under vacuum to give crude product.

This resulting product (0.01 mol) was refluxed for 4-5 hr

(KDB-1 to KDB-9): Equimolar mixture of mol) using hydrochloric acid as catalyst. The completion of with ethanolic solution of different aromatic amines (0.012) reaction was confirmed by TLC using $(7.5:2.5-Hexane)$: Ethyl acetate) mobile phase. After completion of reaction, the reaction mixture was cooled. The resulting solid was filtered, washed with cold ethanol and dried under vacuum to give crude product. The obtained crude product was purified by tituration with diethyl ether. The physical constants of all synthesized compounds are listed in Table 1.

Synthesis of substituted and fused pyrazolopyrimidines (TC-1 to TC-16):

I st Step: Synthesis of substituted acetoacetanilide derivatives (AAA) (Intermediate-I): Equimolar mixture of substituted aromatic amine (I), 1, 3-diketone (II) and catalytic amount of potassium hydroxide (KOH) in 1,

4-dioxane was refluxed for 4-5 hr. The progress of the reaction was monitored by TLC. After completion of reaction, reaction mixture was allowed to cool at room temperature and was poured into crushed ice. The obtained solid was filtered and was purified by titruation with hexane to get pure product (Intermediate-I).

II nd Step: Synthesis of 5-Amino-3-(methylthio)- 1H-pyrazole-4-carbonitrile (Intermediate-II): A mixture of malano nitrile (0.01 m mol) and dry K_2CO_3 (0.012 m) mol) were stirred in dry DMF at room temperature for 30 min., 0.02 mole carbon disulphide (CS_2) was drop wise added in reaction mixture. Then, the reaction mixture was stirred for an additional 2.5 hr at same temperature. The reaction mixture was then cooled at 0-5 °C and dimethyl sulphate (0.02 mol) was added. The solution was stirred at room temperature for another 5-6 hr and was poured into crushed ice to give solid product. The resulting solid was

filtered, washed with cold water and was dried under vacuum to give crude product.

This resulting crude product (0.01 mol) was refluxed with hydrazine hydrate (0.01 mol) for 30 min. in isopropyl alcohol (IPA). After completion of reaction, the reaction mixture was cooled and poured into crushed ice. The resulting solid was filtered, washed with water and dried under vacuum to give crude product (Intermediate-II). The obtained crude product was purified by tituration with hexane and used in next step without further purification.

IIIrd Step: Synthesis of substituted and fused pyrazolo pyrimidines (TC-1 to TC-16): A mixture of Int.-I (0.01 mol), Int.-II (0.015 mol) and different substituted aldehyde (0.01 mol) was heated at 140-150 °C for 25-30 min. in presence 3-4 drops of DMF. The completion of reaction was confirmed by TLC. After completion of reaction, reaction mixture was allowed to cool at room temperature and poured into crushed ice. The resulting solid was filtered, washed with water and dried under vacuum which was then purified by tituration with methanol.

Table 1: Physical constants of2,4-disubsstitutedpyrimidine derivatives.

pyrazolopyrimidines (KD-1 to KD-12): An ethanolic solution of different substituted aromatic aldehyde (0.01 mol), Int.-II (0.015 mol) and malano nitrile (0.01 mol) was refluxed for 3-5 hr using sodium hydroxide as catalyst. The

S**ynthesis of substituted and fused** reaction mass was cooled and resulting solid was filtered, washed with cold ethanol and dried. The crude product was purified by tituration with diethyl ether.

> The physical constants of all synthesized compounds are listed in Table 2.

Table 2: Physical constants of Pyrazolopyrimidine derivatives

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Synthesis of imidazothiazole derivatives (TP-1 to TP-9):

I st Step: Synthesis of ethyl 2-aminothiazole-5 carboxylate (Int.-1): Equimolar mixture of ethyl chloro acetate and ethyl formate were added drop wise to a suspension of 0.01 mol solution of sodium ethoxide in dry toluene, maintained at a temperature between 0-5°C for 2 hr. Then, the reaction mixture was stirred at 0°C for another 2.5 hr. The contents were diluted with water and the layers were separated. The aqueous phase was acidified

with concentrated hydrochloric acid. In this acidified solution, 0.013 mole of aqueous thio urea solution was added and the solution was refluxed for 2.5 hr. The completion of reaction was confirmed by TLC using (100% ethyl acetate) as mobile phase. After completion of reaction, the reaction mass was cooled and neutralized with sodium hydroxide solution. An amber colored solid was precipitated, which was filtered and dried to get desired product ethyl 2-amino thiazole-5-carboxylate.

II nd Step: Synthesis of imidazothiazole derivatives (TP-1 to TP-9): A mixture of ethyl 2-amino thiazole-5-carboxylate (Int.-1) (0.01 mol), different substituted phenacyl bromide (0.012 mol) and 10 % aqueous solution of 1, 4-diazabicyclo [2.2.2]octane (DABCO) was refluxed for 1 hr. The completion of

reaction was confirmed by TLC using (8:2-Hexane: Ethyl acetate) as mobile phase. After completion of reaction, the reaction mixture was cooled. The resulting solid was filtered, washed with cold water and dried. The obtained crude product was purified by tituration with mixture of methanol and ethyl acetate.

Synthesis of fused tetrazolopyrimidines derivatives (K-1 to K-14):

Synthesis of substituted aceto acetanilide derivatives (AAA) (Intermediate-I): Given above (of TC series).

Synthesis of fused tetrazolopyrimidines derivatives

(K-1 to K-14): All the compounds (K-1 to K-14) were synthesized according to synthesis of fused pyrazolopyrimidines (TC series).

The physical constants of all synthesized compounds are listed in Table 4.

The formation of compounds was checked by TLC (Performed on aluminum coated TLC plates gel-G60 $F₂₅₄$ and accomplished on 0.5-mm (E. Merck)). Visualization of spot was made with UV light (254 and 365 nm), an iodine vapor and other visualizing reagent. The melting point was determined in open capillary tubes and was uncorrected. IR spectra were recorded on KBr discs, using FT-IR,

(Shimadzu spectrophotometer Model no.-8400). ¹H-NMR spectra were taken on a Bruker AVANCE II 400. In all the cases, ¹H NMR spectra were obtained in DMSO- d_6 using TMS as an internal standard. The NMR signals are reported in δ ppm. Mass spectra were determined using direct inlet probe on a GCMS-QP-2010 mass spectrometer.

Compound Code	Substitution \boldsymbol{R}	<i>M. F.</i>	M. Wt. (g/mol)	Yield (%)	R_f value
$TP-1$	$4-OCH3$	$C_{15}H_{14}N_2O_3S$	302.35	64	0.41
$TP-2$	$4-C1$	$C14H11ClN2O2S$	306.77	66	0.49
$TP-3$	$4-Br$	$C_{14}H_{11}BrN_2O_2S$	351.22	70	0.48
$TP-4$	$3,4$ -di-F	$C_14H_{10}F_2N_2O_2S$	308.30	61	0.49
$TP-5$	-H	$C_{14}H_{12}N_2O_2S$	272.32	69	0.50
$TP-6$	$4-F$	$C_{14}H_{11}FN_2O_2S$	290.31	72	0.47
$TP-7$	$2,4$ -di-Cl	$C_{14}H_{10}C_{12}N_2O_2S$	341.21	69	0.46
$TP-8$	$4-NO2$	$C_{14}H_{11}N_3O_4S$	317.32	67	0.43
$TP-9$	4 -CH ₃	$C_{15}H_{14}N_2O_2S$	286.35	63	0.48

Table 3: Physical constants of Imidazothiazole derivatives

Biological Screening

The antibacterial and antifungal activities of all synthesized compounds were studied in DMSO. All the synthesized compounds were recrystallized prior to use and DMSO was purified by standard method [27]. For all the compounds, agar well diffusion method was used.

Following Strains were used for the antimicrobial screening:

*Gram positive bacteria***:**

(I) (I)Corynebacterium rubrum ATCC14898 (*CR*) *(II) (II)Staphylococcus albus* NCIM2178 (*SAL*) *(III) (III)Staphylococcus aureus* ATCC25923 (*SA*)

Table 4: Physical constants of Tetrazolopyrimidine derivatives

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*Gram negative bacteria***:**

(I)Enterobacter aerogenes ATCC13048 (*EA*) *(II)Escherichia coli* NCIM2931 (*EC*) *(III)Salmonella typhimurium* ATCC23564 (*ST*)

*Fungi(Yeast)***:**

(I)Candida albicans ATCC2091 (*CA*) *(II)Candida neoformans* NCIM3542 (*CN*) *(III)Candida glabrata* NCIM3448 (*CG)*

All these strains were obtained from National Chemical Laboratory (NCL), Pune, India. The bacterial and fungal strains were maintained on nutrient agar and MGYP medium (Hi Media, India) respectively while *E. coli* were maintained on Luria medium (Hi Media, India) at 4°C and sub cultured before use.

MIC refers to the lowest concentration of the antimicrobial agent which is required for the inhibition of visible growth of the tested microorganism [28]. The antimicrobial activity of an agent is usually quantified by determining the MIC values which serve as a guide for treatment of most infections. MIC values were calculated using INT dye. The MBC is interpreted as the lowest microorganisms.

Preparation of solution of compounds for MIC and MBC study: All the compounds dissolved in DMSO were first diluted to highest concentration $(20 \text{ mg } \text{m} \text{m}^{-1})$ to be tested and then serial two-fold dilution was made in a concentration range from $(0.156 \text{ to } 20 \text{ mg ml}^{-1})$.

Preparation of bacterial inocula for MIC and MBC study: The inocula of the test organisms were prepared using the colony suspension method [29]. Colonies picked from 24 h old cultures grown on nutrient agar were used to make suspension of the test organisms in saline solution to give an optical density of approximately 0.1 at 600 nm. The suspension was then diluted 1:100 by transfer of 0.1 ml of the bacterial suspension to 9.9 ml of sterile nutrient broth before use to yield 6×10^5 CFU ml⁻¹. .

concentration that can completely remove the for 24 h**.** Experiments were carried out in duplicate. After -1) to be well [31]. The plates were incubated for further 30 min and **Determination of the minimum inhibitory concentrations (MIC):** The MIC was determined by the micro well dilution method [30] with some modification. This test was performed in sterile flat bottom micro test plates (Tarsons Products Pvt. Ltd.). 150 µl volume of Mueller Hinton broth (MHB) was dispensed into each well and 20 µl of various concentrations of the compounds was added in decreasing order along with 30 µl of the test organism suspension. The final volume in each well was 200 μl (150 μl Mueller Hinton broth, 30 µl of the test organism suspension and 20 μl compound). Two control wells were maintained for each test batch; sterility control (MHB and DMSO) and organism control (MHB, test organism and DMSO). Plates were then incubated at 37°C incubation, 40 µl of INT (2-(4-Iodo phenyl)-3-(4-nitro phenyl)-5-phenyltetrazolium chloride) solution (0.2 mg ml⁻¹) dissolved in sterile distilled water was added to each were estimated visually for change in color to pink indicating reduction of the dye due to bacterial growth. The

highest dilution (lowest concentration) that remained clear corresponded to the MIC.

concentration (MBC): MBC was determined from all wells showing no growth as well as from the lowest concentration showing growth in the MIC assay for all the samples. Bacterial cells from the MIC test plate were sub cultured on freshly prepared solid nutrient agar plates by making streaks on the surface of the agar. The plates were incubated at 37°C for 24 h overnight. Plates that did not show growth were considered to be the MBC for the compounds used [32]. The experiment was carried out in duplicate.

RESULTS AND DISCUSSION

2, 4-disubstituted pyrimidine derivatives (BKD and KDB series):

The MIC and MBC values of BKD compounds are presented in Table 5. The compounds exhibited concentration dependent inhibition of growth. All the compounds showed varied levels of MIC and MBC values against studied microorganism. In sterility control (MBH and DMSO), DMSO had no inhibitory effect on the tested organisms. For the Gram positive bacterial strains MIC and MBC varied from ≤ 0.156 mg ml⁻¹ to ≥ 20 mg ml⁻¹ and 0.250 bac mg ml⁻¹ to >20 mg ml⁻¹ respectively for BKD series S. albus is most resista whereas for KDB series, MIC and MBC varied from ≤ 0.156 mg ml⁻¹ to >20 mg ml⁻¹ and 10 mg ml⁻¹ to >20 mg ml ml⁻¹ respectively.

Against *S. albus,* all compounds showed MIC and MBC values >20 mg ml⁻¹. Against *S. aureus, compound BKD-4* 4-CF₃ and 3-Cl are and BKD-6 showed lowest MIC $\left($ < 0.156 mg ml⁻¹) whereas maximum is obseved by BKD-7 having value of 10 mg ml⁻¹. The lowest MBC value is 0.250 mg ml⁻¹ for BKD-6 KDB-9. followed by BKD-4 and BKD-9 (having value 10 mg ml⁻¹). BKD-10 and BKD-12 had minimum MIC value of 0.625 mg ml⁻¹ against *C. rubrum* whereas MBC values were 20 MIC values aga mg ml⁻¹ for all the studied BKD compounds. t_1

For all the selected Gram negative bacterial strains, MBC values are >20 for all the BKD compounds. The MIC values are minimum i.e., 5 mg ml⁻¹ for BKD-1, BKD-2 and mg ml⁻¹ respecti KDB-10 against *E. aerogenes.* For *E. coli also, B*KD-2 has minimum value of 5 mg ml⁻¹ whereas a value of 5 mg ml⁻¹ -1 is for BKD-10 against *S. typhimuriu* which is minimum as comparison to other compounds.

Determination of the minimum bactericidal minimum value of MIC is for BKD-1 and maximum is for For different fungal strains, all the compounds showed varied levels ofMIC and MBC values. Against *C. albicans,* BKD-12. MBC values were >20 mg ml⁻¹ for all the studied BKD compounds. The lowest MIC values $(\leq 0.156 \text{ mg ml}^{-1})$ shown by BKD-1 (against *C. glabrara)* and BKD-3 and BKD-5 against *C. neoformans*. The minimum MBC values are 10 mg ml⁻¹ and 5 mg ml⁻¹ respectively for these two fungal strains.

> -1 and 0.250 bacteria are most resistant. Among Gram positive bacteria, The inhibition depends on solvent, compound structure and strain. In the present study, solvent is same throught so this parameter is not considered. Table 1 shows that different R groups in these compounds. These compounds have same central nucleus but different substitution. These substitutions are aryl ring with different functional groups. Thus, different substitutions affect different strains differently. In BKD series, BKD-4 and BKD-6 contains 3 -CF₃ and 4 -OCH₃ respectively. Thus, 3 -CF₃ and -OCH₃ substitutions are more effective against*S. aureus*. Whereas 4-chloro present in compound BKD-1 is more effective against fungal strain *C. glabrara*. Against *C. neoformans,* BKD-3 and BKD-5 compounds are more effective containing 4-fluoro and 3-chloro, 4-fluoro respectively. For Gram negative bacteria, the studied BKD compounds are not very effective. Thus, these selected Gram negative *S. albus* is most resistant.

to >20 mg ml -1 value of MIC against *S. albus*. However, *S. aureus,* ⁻¹) whereas values were >20 mg ml⁻¹ for all the studied KDB Table 6 shows that all the KDB compounds have >20 mg KDB-4 and KDB-8 showed lowest MIC $(\leq 0.156 \text{ mg ml}^{-1})$ and MBC (10 mg ml⁻¹) values. Thus, for this bacteria, 4-CF³ and 3-Cl are most effective. For *C. rubrum,* MBC compounds but MIC values are minimum (2.5 mg ml^{-1}) for KDB-9.

-1). Thus, 2-flouro substitution is most effective for this strain. All the compounds showed varied and moderate MIC values against *E. coli, E. aerogenes* and *S. typhimurium.*However, MBC values are >20 for all the compounds. For the fungal strains, MIC and MBC varied from ≤ 0.156 mg ml⁻¹ to ≥ 20 mg ml⁻¹ and 5 mg ml⁻¹ to ≥ 20 mg ml -1 respectively. Minimum MIC is for KDB-9 containing 2-flouro against *C. neoformans.*

Fused pyrimidine derivatives (TC, KD and K series):

Table 7 shows the MIC and MBC values of TC series.

TC-9 showed minimum value of MIC $(>0.156$ mg ml⁻¹) followed by TC-1 (>0.425 mg ml⁻¹) against *S. albus.* compo Whereas TC-5, TC-7, TC-8 and TC-10 compounds are more effective against*S. aureu*. For *C. rubrum*, TC-2 and TC-15 showed minimum MIC value of 1.25 mg ml⁻¹. The Again MBC values are >20 mg ml⁻¹ for all the compounds against chloro, *S. albus* and *C. rubrum*. Only TC-7, TC-8 and TC-15 had MBC value of 10 mg m l^{-1} . .

As shown in Table 2, TC compounds have same central nucleus but different R_1 , R_2 and R_3 groups. TC-9 contains 3, 4-di chloro, (CH3)2CH- and 4-OCH³ whereas TC-1 contains 4- chloro, $(CH_3)_2CH$ - and 4- chloro groups at R_1 , R_2 and R_3 positions respectively. Thus, it is observed that when 4- chloro and 4-OCH₃ groups are present at R_3 MIC

-1) is minimum. In TC-2, TC-5, TC-7, TC-8, TC-10 and TC-15 2 and *aureu* and *C. rubrum*.
Against Gram negative bacterial strains, when R₁ is 3compounds also, 4-chloro and 4-OCH³ groups are present at R³ which causes a decrease in MIC values against *S.*

chloro, $4-F$ and $4-OCH_3$ groups, compounds are more effective against*E. aerogenes*. TC-13 containing 4-chloro (R_1) , $-CH_3$ (R_2) and $4-Br$ (R_3) shows lowest MIC value against *E. coli* whereas against *S. typhimurium,* not a single compound was effective. Against *E. coli* and *S. typhimurium*, all the compounds have MBC values >20 mg ml⁻¹. Only TC-2 has lowest MBC of 0.300 mg ml⁻¹ against *E.aerogenes*.

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Table 10: Antimicrobial activity data (MIC and MBC in mg ml ⁻¹)								of TP series compounds against Gram positive bacteria, Gram negative bacteria and fungal strains.										
				Gram positive bacteria ⁺						Gram negative bacteria+						Fungi(yeast)		
Conwd.		$\mathcal{S}4L$		$SA +$		\mathbb{R}^2		E4₽		EC		$ST^{\scriptscriptstyle\downarrow}$		Ĝ₽		ري		CN.
$Code^{\circ}$		MIC MBC MIC MBC MIC MBC						MIC MBC				MICA MBCAMICA MBCAMICA MBC				MIC MBC	MIC MBC	
$TP - I = 20 + 20 = 1$						10ω >20 ω >20 ω >20 ω	>204	>200 - >200 - >200			>204	$>20+20+20+20+$			>204 >204		>204	>20
$TP-2+$	54	>204	$10\div$	>204	54	>20	54	>204	>204	>20	2.5φ	>204	>204	>20	>204	>20	>20	>204
$TP-3+$	54	>20	54	>20	2.5ω	>20	54	>204	>204	>20	1.254	>204	>204	>204	>20	>204	>204	>204
$TP-4$	>204		>200 ⁻¹ >200 ⁻¹	>20	204	>204	>204	>204	>204	>204	>20	>204	>204	>204	>204	>204	>204	>204
$TP-5^{\circ}$	>204		>200 >200	>204	>204	>20	>204	>204	>204	>20	>204	>204	>204	>204	>204	>204	>204	>204
$TP\text{-}6$ e $>$ 20e			>200 >200	>204	204	>204	54	>20	>204	>20	>204	>204	>204	>204	>204	>204	>20	>204
$TP - 74 = 204$ > 204 2.54 > 204 1.254 1.254							54	>20	>204	>20	54	>204	2.5ω	54	$1.25*$	54	>204	>20
$TP-8-$		204 >204 2.54 >204			$2.5\div$	$>20\div$	2.54		>204 >204	>20	1.54	>204	>204		>204 >204	>204	>204	>204
TP.94		$20 - 20 - 20 - 20 - 20 - 20 - 20 - 20$				>204		>2041 >2041 >2041 >2041 >2041 >2041 >2041 >2041 >2041 >2041 >204										>204

position in compound TC-7 against *C. glabrata* and TC-7 and TC-15 for *C. neoformans* have minimum MIC values. Comparison of different groups at R_1 , R_2 and R_3 positions shows that when 4-F group is at R_1 position as in TC-4 and TC-15, it is more effective.

This is followed by 4 -CF₃ (as in TC-7) and 4 - chloro (as in $TC-11$) at R_1 position. In all these four effective compounds, R_3 is 4- chloro or 4-OCH₃. The R_2 position is found to be not very effective. However, all these TC compounds have >20 mg ml⁻¹ MBC values.

Table 8 shows MIC and MBC values of KD series against different bacterial and fungal strains. For this series also, MBC values are not very significant against all the bacterial strains. However, among the three fungal strains, against *C. glabrata,* KD-7 to KD-12 compounds has MBC value of 10 mg ml⁻¹. For other two strains, values are \geq 20 compounds exhib mg ml -1 for all the compounds.

KD-4, KD-6, KD-9, KD-11 and KD-12 compounds show lowest MIC values against Gram positive bacterial strains. For fungal strains, compounds KD-3, KD-6, KD-7 and KD-12 are most effective. However, all the compounds had little effect against Gram negative bacterial strains.

Table 2 shows the general structure of these derivatives along with different R groups. KD-4 and KD-9 containing 4-F and 3-OCH³ groups, affect *S. albus*. Against *S. aureus*, KD-6, KD-11 and KD-12 containing 3- chloro, -4CH₃ and 3, 4, 5-tri OCH³ groups are found to be most effective.

Against Gram negative bacteria, overall methoxy and chloro groups at different positions are found to be a little bit effective.

Against fungal strains, 4- chloro and 3-Br containing compounds KD-3 and KD-7 are effective against C. *albicans*. However, *C. glabrata* is most inhibited by compound KD-6, KD-7 and KD-12 containing 3-Cl, 3-Br and 3,4,5-tri OCH³ respectively. Against *C. neoformans,* some compounds show minimum value of MIC to be 2.5 mg ml⁻¹. Thus, for KD series, the selected Gram negative strains. The MBC va bacteraia are most resistant. Among the studied Gram positive bacteria and fungal strains, *C. rubrum* and *C. neoformans* are most resistant for this series.

Table 9 shows the antimicrobial activity data of tetrazolopyrimidine derivatives $(K-1)$ to $K-14$ against bacterial as well as fungal strains. It is evident from Table 4.5 that against *S. albus*, MIC value is 5 mg ml⁻¹ for K-13 which is lowest as comparison to other compounds.

Table 2 shows the different substitutions in these

compounds. Thus, different substitution affect different strain differently. So, 2-OCH³ group present in K-13, is most effective MIC values are minimum for K-2 and K-6 against *S. aureus*. Both these compounds again contain methoxy groups at different positions. Thus, methoxy group is most effective against *S. aureus* also. For *C. rubrum*, 2.5 mg ml⁻¹ MIC is found to for few compounds; K-9,K-12 and K-14. Thus, for this bacterial strain, 3-Br, 2, 5-di OCH³ and 2-Cl groups are found to be most effective.

Against different Gram negative bacterial strains, K-4 and K-11 compounds are more effective against *E. aerogenes*. Other compounds had no signifacnt effect against *E. coli* and *S. typhimurium*. Thus, 4-F and 3-OCH³ are most effective against *E. aerogenes.*

Against fungal strains, compounds, only K-3 and K-11 compounds exhibited minimum MIC value of ≤ 0.156 against *C. neoformans* and *C. glabrata* respectively. K-3 also has lowest MIC values (0.310 mg ml⁻¹) against *C*. *albicans*. Thus, 4-chloro (as in K-3) and 3-OCH³ (as in K-11) groups are most effective against *C. neoformans* and *C. glabrata* respectively.

Table 10 shows the MIC and MBC values of imidazothiazole derivatives (TP-1 to TP-9) against nine bacterial and fungal strains. It is observed that among the three Gram positive bacteria, TP-2 and TP-3 against *S.* albus and TP-7 and TP-8 compounds (2.5 mg ml⁻¹) against *S. aureus* showed lowest MIC as compared to other compounds. Against *C. rubrum*, TP-7 has minimum MIC of 1.25 mg ml⁻¹. The general structure of these compounds of TP series along with different substitutions (R) are given in Fig. 4.5. Thus, 4-chloro and 4-bromo are most effective against *S. albus*, 2.4-di chloro and 4-NO₂ against *S. aureus* and 2,4-di chloro against *C. rubrum* gives better results. In case of Gram negative bacteria, not a single compound gave significant MIC against *E. coli.* For Whereas TP-7 compound also exhibited better results for two fungal strains. The MBC values are >20 mg ml⁻¹ for all the compounds against *S. albus* and *S. aureus*. However, TP-7 having 2, 4-dichloro substitution, exhibited lowest MBC against *C. rubrum*.

-1 for K-13 and TP-3 also show lowest MIC values (2.5 and 1.25 mg In Gram negative bacterial strains, TP-8 having 4-NO₂ shows the lowest MIC value (2.5 mg ml^{-1}) against *E*. *aerogenes* and *S. typhimurium*. For *S. typhimurium,* TP-2 ml⁻¹). Thus, again 4-Chloro and 4-bromo groups are most effective. For *E. coli*, not a single compound has minimum

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Against fungal strains, onlt TP-7 containing 2,4-dichloro groups exhibited better results of MIC and MBC against*C. albicans* and *C. glabrata* fungal strains. Other compounds have $>$ 20 mg ml⁻¹ values for both MIC and MBC against bi and tricyclic pyrim these strains. For *C. neoformans*, these TP series compounds are not effective.

Thus, among the studied bacterial and fungal strains, *E. coli* and *C. neoformans* are most resistant for this series.

Thus, it is concluded that some of the studied compounds in different series can be used as a lead molecule for further biological study, since these compounds exhibited better activity against different strains.

Confilict of Interest

None

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